# Normal Modes of the $MoO_4^{2-}$ lon in $Tb_{1.8}Eu_{0.2}(MoO_4)_3$ Single Crystal

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Received July 22, 1982; in revised form December 7, 1982

Polarized Raman spectra (single crystal) at 300 K and infrared spectra (powder) at 300 and 77 K in the region 250–1000 cm<sup>-1</sup> of a binary molybdate of terbium and europium have been recorded. Based on  $C_{2v}$  symmetry, group theoretical analysis has been carried out and a vibrational assignment is proposed.

77

### Introduction

Rare earth molybdates with chemical formula  $R_2(MoO_4)_3$ , where R is a rare earth element, crystallize in different structures (1, 2). Of these, the molybdates with R =Sm, Eu, Gd, Tb, and Dy are isotypic and crystallize in an orthorhombic Pba2 structure at room temperature and transform to tetragonal  $P42_1m$  system above the Curie temperature (3, 4). They exhibit ferroelectric (5) and ferroelastic (6) properties below the Curie temperature ( $T_c \sim 160-180^{\circ}$ C). Binary molybdates of Sm, Eu, Gd, Tb, and Dy with varying proportions are found to crystallize in the same orthorhombic Pba2 structure (3, 4). Tb<sub>1.8</sub>Eu<sub>0.2</sub>(MoO<sub>4</sub>)<sub>3</sub> (hereinafter termed TEMO) is a binary molybdate in which 10 at.% of Eu<sup>3+</sup> ion replaces the  $Tb^{3+}$  ion in the  $Tb_2(MoO_4)_3$  structure (3).

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dolinium, terbium, and samarium molybdates have been reported (7-10). The soft mode (8) and temperature dependant Raman scattering studies on gadolinium molybdate have also been investigated in detail (7, 11). Koningstein and Predhomme (12) have studied the Raman spectrum of terbium molybdate single crystal in the region 750-1000  $cm^{-1}$  in an attempt to identify its structural similarity with gadolinium molybdate and also to know the site symmetry of the  $MoO_4^{2-}$  ions. Polarized ir and Raman spectra of double molybdates of the type  $KLn(MoO_4)_2$  where Ln = lanthanide ion have been reported recently (13, 14). Very recently we have investigated the infrared and Raman spectra of gel grown lanthanum molybdate (15). A vibrational analysis of TEMO is expected to give information about the distribution of the vibrational energy levels of  $MoO_4^{2-}$  ions in an orthorhom-

Vibrational spectroscopic studies of ga-

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bic lattice. In the present investigation the Raman spectra (single crystal) at 300 K and the ir spectra (in powder form) at 300 and 77 K of TEMO have been reported. The observed spectra have been interpreted using factor group analysis.

## Experimental

Melt-grown optically good quality single crystal of size  $5 \times 6 \times 0.5$  mm was used for the investigation. X-ray and polarizing microscopes were used for the identification of the crystallographic axes. The crystals are cleavable perpendicular to the growth axis z. The ir spectrum was recorded on a Perkin Elmer 580 spectrophotometer at 300 and 77 K. A Specac low temperature cell with liquid nitrogen as the coolant was employed. A Spex Ramalog 1403 spectrometer equipped with a Spectra-Physics model 165 argon ion laser was used to obtain the Raman spectra. The spectra were obtained in the Stokes region of the green line at 5145 Å. The notation of Damen et al. (16) was used for describing the experimental configuration.

#### **Factor Group Analysis**

The binary molybdate TEMO is isotypic with terbium molybdate, crystallizes in an orthorhombic  $Pba2(C_{2v}^8)$  structure at room

temperature (17) and has four molecular units in the primitive unit cell. A crystallographic projection of the elementary unit cell is shown in Fig. 1. The primitive cell contains 12 molybdenum atoms surrounded by oxygen tetrahedra. There are three crystallographically distinct  $MoO_4^{2-}$  groups. The Tb and Eu atoms have 7 oxygen atoms as nearest neighbors. All the Mo atoms are in general positions. This leads to a decrease in the overall symmetry of the  $MoO_4^{2-}$  ion from  $T_d$  to  $C_1$ . The factor group analysis can be done either by correlating the site symmetry of  $MoO_4^{2-}$  ions in  $C_1$  sites to the factor group or by correlating the site group of each atom in the primitive unit cell individually to the factor group. Both will give the same total number of modes for each symmetry species.

The correlation table for the different normal modes of the  $MOQ_4^{2-}$  ion between the free ion symmetry  $T_d$ , the site symmetry  $C_1$  and the unit cell symmetry  $C_{2v}$  is shown in Table I. In this case all the species become ir and Raman active. Considering the whole unit cell, each vibration splits into 12 modes,  $9(3A_1 + 3B_1 + 3B_2)$  of them being Raman and ir active and the other  $3(3A_2)$  Raman active only. The analysis, based on the line splitting of the normal modes of free  $XY_4$  ion is expected to provide a straightforward means for identifying the different normal modes.

Free ion $T_d$			Site $C_1$			Factor group $C_{2v}$	
Mode	Vibration	Activity <sup>a</sup>	Mode	Activity <sup>a</sup>	Mode	Raman Tensor	Activity <sup>a</sup>
$\nu_1 A_1$	v <sub>s</sub> (MoO <sub>4</sub> )	R			27 <b>A</b> 1	$\alpha_{xx}, \alpha_{yy}, \alpha_{zz}$	ir, R
$\nu_2 E$	$\nu_{\rm s}({\rm MoO_4})$	R			27A2	$\alpha_{xy}$	R
			9C1	ir, R			
$\nu_3 F_2$	$\nu_{\rm as}({\rm MoO_4})$	ir, R			$27B_{1}$	$\alpha_{zx}$	ir, R
$\nu_4 F_2$	$\nu_{\rm as}({\rm MoO_4})$	ir, R			$27B_{2}$	$\alpha_{zy}$	ir, R

TABLE I Correlation Table and Selection Rules for the  $MoO_4^{2-}$  Ion in TEMO Crystal

<sup>a</sup> R, Raman; ir, infrared.

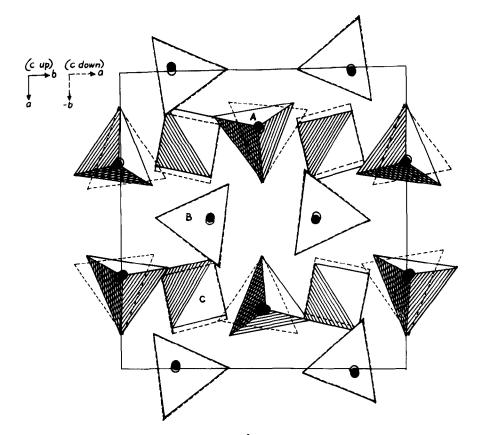


FIG. 1. Arrangement of  $RE^{3+}$  (circles) and  $MOQ_4^{2-}$  ions in the TEMO unit cell. Solid lines represent atomic positions in a cell with orientation a b c and dashed lines are those in a cell with orientation -b, a - c. (after Ref. 17). A, B, C are the three crystallographically different MoO<sub>4</sub> tetrahedra.

#### **Results and Discussion**

Müller *et al.* (18) reviewed the numerous studies regarding the distribution of vibrational levels of the  $MoO_4^{2-}$  ion in different normal molybdates. A discrepancy existed as to whether the symmetric bending  $\nu_2 >$ the asymmetric bending  $\nu_4$ , or  $\nu_4 > \nu_2$ . The intensity studies on single crystal data and the comparative study of the various potential functions for the MoO<sub>4</sub> species conclusively proved that  $\nu_4 > \nu_2$ . The interpretation of the vibrational spectra is often difficult when the primitive unit cell becomes complex with fairly large number of molybdate ions. The situation is further complicated when the molybdenum-oxygen coordination forms layered type compounds.

The normal modes of the molybdate ion are distinguishable from the external modes, since their intensities are an order of magnitude higher and bandwidths are narrower by at least a factor of three. (The spectra has been recorded in the region  $250-50 \text{ cm}^{-1}$  which also gives the external modes.) The normal modes of the molybdate ion are expected to occur in the regions 750–975 cm<sup>-1</sup> (stretching) and 250– 450 cm<sup>-1</sup> (bending) irrespective of its environment.

## Raman Spectra

Raman spectra of TEMO single crystal in the region 250–1000 cm<sup>-1</sup> are obtained for the five orientations y(xx)z, x(yy)z, x(yx)z, y(zx)z, and y(zy)z (Fig. 1). Attempts to record the y(zz)x spectra were unsuccessful. Observed lines are tabulated in Table II. The symmetries of the Raman active phonons that can participate in different orientations can be had from the matrix representations of the Raman tensors for the appropriate crystallographic point groups given by Loudon (19). For orthorhombic crystals the polarizability tensor for the  $A_1$  class is

$$\begin{bmatrix} a & 0 & 0 \\ 0 & b & 0 \\ 0 & 0 & c \end{bmatrix} \quad a = \frac{\partial \alpha_{xx}}{\partial r_{\mu}},$$

$$b = \frac{\partial \alpha_{yy}}{\partial r_{\mu}}, \quad c = \frac{\partial \alpha_{zz}}{\partial r_{\mu}}.$$

This indicates that the particular lines in  $A_1[(xx), (yy), \text{ and } (zz)]$  should differ in intensity which is found to be true in the case of xx and yy spectra (Fig. 2).

The observed spectra is expected to be very complex with 108 modes (Table I) as the primitive unit cell contains 12 molyb-

TABLE II

Frequencies, Occurring Orientations, Intensities<sup>a</sup> and Assignments in Raman and Observed Frequencies at 300 K and 77 K in Infrared Spectra of TEMO Crystal

	x(yy)z A <sub>1</sub>	x(yx)z $A_2$	y(zx)z B1		Infrared frequencies at	
y(xx)z $A_1$				y(zy)z $B_2$	300 K	77 K
960 (750)	960 (560)	960 (560)	960 (120)	960 (275)		
( )	· · · ·					956, 950
942 (860)	942 (660)	942 (610)	942 (60)	942 (180)	943	943
. ,		920 Sh	919 Sh	917 Sh	915	915
905 (20)	905 (20)	902 Sh				
	851 (390)					
			850 (150)			
848 (330)		848 (870)		848 (360)		856, 850
					845	
			842 (150)			846, 840
					824	824, 810
838 (100)						807
825 (80)						
	821 (140)			821 (1000)		
		820 (205)	820 (800)			
819 Sh						
		806 (90)				
	803 (80)					
801 (90)						
		762 (30)		7/1 51		
	750 (25)			761 Sh		
756 (20)	759 (25)				745	745
730 (20)				743 (460)	745	743
			742 (205)	743 (400)		
		738 (85)	772 (203)			
	735 (45)	/30 (03)				

<i>.</i> .	x(yy)z $A_1$	x(yx)z $A_2$	y(zx)z $B_1$	y(zy)z $B_2$	Infrared frequencies at	
y(xx)z $A_1$					300 K	77 K
·	•	-		-		
734 (40)					720 Sh	720 Sh
			718 Sh	717 Sh	720 511	720 511
	712 Sh	713 Sh				
449 (90)	449 (35)	449 (50)	449 (40)	449 (30)		
			(10 (35)	415 (80)	415 w,Sh	(10
			410 (25)		410 395 s,Sh	410 395 w,Sh
				389 (60)	575 8,511	575 w,311
	388 (310)			(,		
386 (590)			386 (65)			
		383 (1020)				
364 (540)	362 (340)					
	302 (340)	357 (420)				356
		557 (120)			352	550
		347 Sh				
			345 Sh	348 (390)		
333 Sh				220 (090)	340	340 330 Sh
333 Sn			329 (1240)	330 (980)	330	330 Sn
	326 (500)		525 (1240)			
324 (890)	<b>、</b>				325	
		323 (840)				
316 Sh				210 (170)		
				310 (170)	305	305
			306 (295)		505	505
302 (70)			,			
	300 (70)					
		297 (130)		<b>201</b> (100)		
288 (80)	288 (50)		288 (240)	291 (180)		286
200 (00)	200 (50)	283 (140)	200 (270)			200
					282	

TABLE II—Continued

Note. w = weak, s = strong, Sh = Shoulder.

<sup>a</sup> Intensity values given in brackets not normalized.

date ions. All the predicted modes could not be observed. The oxygen environment of the 3 crystallographically different molybdenum atoms is almost tetrahedral with average Mo-O distances 1.762, 1.761, and 1.753 Å (17). A calculation of the phonon energies for the Mo-O stretching and bending vibrations, making use of the different Mo-O distances is expected to give closely spaced phonon energies, which has been the case for gadolinium molybdate (20). Even with the LO-TO recordings of gadolinium molybdate single crystal at 77 K all the predicted modes could not be observed (11b).

The two strong lines at 960 and 942 cm<sup>-1</sup>

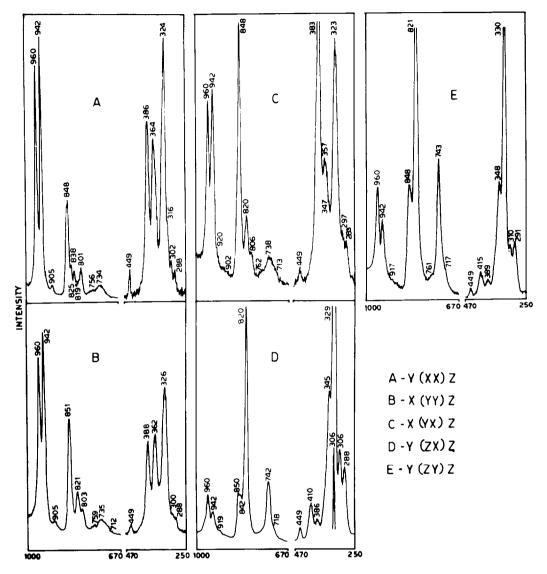


FIG. 2. Raman spectra of TEMO single crystal.

(xx and yy spectra) are assigned to the totally symmetric vibrations of  $MoO_4$  groups. Factor group splittings are not observed for these lines. Theoretically there should be 12 symmetric stretchings since there are 12  $MoO_4$  units present in the primitive unit cell. These 12  $MoO_4$  units form 3 sets of 4 crystallographically different  $MoO_4$  units with average Mo-O distances of 1.762, 1.761, and 1.753 Å. Even if the 3 sets are vibrating with the same energy, there should be 3 lines for the symmetric stretching in the observed spectra. The presence of 2 lines can only be justified by assuming that there is an accidental degeneracy between 2 sets of 4 MoO<sub>4</sub> units with average Mo–O distances 1.762 and 1.761 Å. Hence the line at 942 cm<sup>-1</sup> is assigned to these 2 sets of 4 MoO<sub>4</sub> units and the 960 cm<sup>-1</sup> to the third set with the average Mo–O distance 1.753 Å. Further, the intensity distribution of these 2 lines (intensity of 942 cm<sup>-1</sup> > intensity of 960 cm<sup>-1</sup>) supports our reasoning. A low temperature measurement of LO-TO splittings is expected to resolve these lines.

Out of the 36 asymmetric stretchings predicted, only 25 could be observed (Table II). These lines fall in the range 712–917  $cm^{-1}$ . The observed splitting of 205  $cm^{-1}$ for the  $\nu_3$  stretchings is fairly large. There are lines with a difference of 1 or 2  $cm^{-1}$  if the different orientations are taken together. For example, the 848  $cm^{-1}$  in (yx) spectra and 850  $cm^{-1}$  in (zx) spectra cannot be considered as "leakage" (from misalignment and the effect of collecting angle of the focusing lens) in view of the fact that the phonon energies are very closely spaced. This point is further confirmed by the lines at 848 cm<sup>-1</sup> (xx spectra) and 850 cm<sup>-1</sup> (yy spectra). The intensity distribution of these lines with respect to the 960 or 942 cm<sup>-1</sup> lines is also very much different.

From intensity considerations, Müller et al. (18, 21) have shown that  $\nu_2$  has a higher intensity than  $\nu_4$  in Raman and the reverse in infrared. This condition along with the earlier stated one that  $\nu_4 > \nu_2$  have been used in the assignment of  $\nu_2$  and  $\nu_4$ . The lines at 386, 364 cm<sup>-1</sup> (xx); 388, 362 cm<sup>-1</sup> (yy) are assigned to  $\nu_2$  of the different molybdate ions. The strong line at 324 cm<sup>-1</sup> (xx) with factor group splittings at 316 and  $302 \text{ cm}^{-1}$  is also assigned to  $\nu_2$ . The following lines: 383, 357, 297, and 283 cm<sup>-1</sup> (yx) spectra; 410, 345, 329, 306, and 288 cm<sup>-1</sup> (zx) spectra; 415, 348, 330, 310, and 291  $cm^{-1}$  (zy) spectra are assigned to  $v_4$ . The present assignment differs from the one due

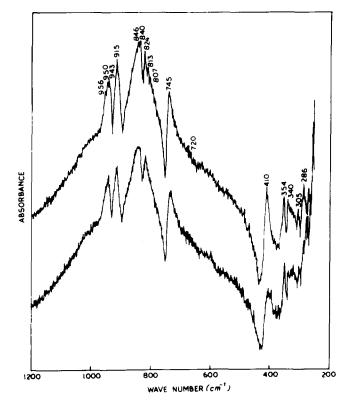


FIG. 3. Infrared spectra of TEMO (KBr pellet).

to Ullman *et al.* (7) which is based on the assumption  $\nu_2 > \nu_4$ . It may be noted that the labeling of  $\nu_2$  and  $\nu_4$  as internal modes is only approximate since mixing is expected with lattice modes.

#### Infrared Spectra

The room temperature and the liquid nitrogen temperature spectra are shown in Fig. 3. The spectra at 77 K shows some splittings and an enhancement in intensity which is pronounced in the  $\nu_2$  and  $\nu_4$  regions of the molybdate ions. All the  $\nu_1$  and  $\nu_3$ modes fall in the broad contour from 600 to 1000 cm<sup>-1</sup>. The 943 cm<sup>-1</sup> band at 300 K shows splittings at 950 and 956 cm<sup>-1</sup> when the temperature is lowered. These are assigned to the symmetric stretching vibration of molybdate ions. The symmetric nature of the band at 915 cm<sup>-1</sup> warrants an assignment of  $v_1$  for this. However, there is no corresponding Raman line in the xx or yy spectra. An assignment of  $\nu_1$  for 915 cm<sup>-1</sup> tends to increase the difference in frequency between the crystallographically different molybdate ions, which is not expected since all these ions are almost tetrahedral and their average Mo-O distances are very close. Further, the shoulder at 917  $cm^{-1}$  in the zx geometry in Raman has been assigned to  $v_3$ . Hence, the band at 915 cm<sup>-1</sup> has to be taken as a  $\nu_3$  component. The broadbands centered at 845 cm<sup>-1</sup> and 824  $cm^{-1}$  at 300 K with splittings at 850, 846, 840, 824, 810, and 807 cm<sup>-1</sup> at 77 K also belong to  $\nu_3$ . The highly asymmetric band at 745  $cm^{-1}$  with a shoulder at 720  $cm^{-1}$  has to be associated with an asymmetric stretching. The intensity rule (18, 21) dictates that  $\nu_3$  has a higher intensity than  $\nu_1$  in infrared and the reverse in Raman holds good in our study.

The bending region ( $\nu_2$  and  $\nu_4$ ) shows frequencies corresponding to that observed in Raman. There are three band contours. The first one centered at 410 cm<sup>-1</sup> has a weak shoulder at 415 cm<sup>-1</sup> and a strong one at 395 cm<sup>-1</sup>. Nearing 77 K this band gains in intensity and in the process the weak shoulder at 415  $cm^{-1}$  disappears and the strong one is observed as a weak but prominent one at 395 cm<sup>-1</sup>. In conformity with the intensity rule and Raman assignment, the lines at 415 and 410 cm<sup>-1</sup> have been assigned to  $v_4$  and the line at 395 cm<sup>-1</sup> to  $\nu_2$ . The second band contour extends from 300 to 370 cm<sup>-1</sup> with peaks at 352, 340, 330, and 325  $cm^{-1}$  at room temperature. At 77 K the 352 cm<sup>-1</sup> shifts to 356 cm<sup>-1</sup> without gaining intensity and the asymmetric nature is more pronounced on the higher wavenumber side. This band is assigned to a  $\nu_4$  mode. The line at 340 cm<sup>-1</sup> gains some intensity and during the process the lines at 325 and 330  $cm^{-1}$ coalesce, forming a shoulder at 330  $cm^{-1}$ . The 340 and 325 cm<sup>-1</sup> are assigned to  $\nu_2$  and the 330 cm<sup>-1</sup> to  $\nu_4$ . The line at 305 at both the temperatures belongs to a  $\nu_2$  component. The third band contour centered at 282 cm<sup>-1</sup> at 300 K shifts to 286 cm<sup>-1</sup> at 77 K. This has been assigned to a  $v_4$  component.

To conclude, the site approximation without the inclusion of factor group effect is insufficient to interpret the large number of frequencies observed in the spectra. The Mo-O stretching and bending frequencies occupy a wide range of the spectrum indicative of the large number of molybdate ions in the primitive unit cell.

#### Acknowledgments

The authors are grateful to Professor L. H. Brixner, Delaware, for providing the crystal, and to Mr. M. Pal, for his assistance in recording the spectra. One of the authors (SSS) is thankful to the CSIR for the award of a fellowship.

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